Course Title:	Introduction to Nano Technology									
Course Code:		BETCK105C/205C	CIE Marks	50						
Course Type (Theory/Practical		ETC (Integrated)	SEE Marks	50						
/Integrated)			Total Marks	100						
Teaching Hours/Week (L:T:P: S)		02:00:02:00	Exam Hours	03						
Total Hours of H	Pedagogy	40 hours	Credits	03						
Teaching Depar	tment	NT/Chem/Phys/Any Engg. Branch	ys/Any Engg. Branch QP setting N							

Course objectives

- To provide a comprehensive overview of synthesis and characterization of nanoparticles, nanocomposites and hierarchical materials with nanoscale features.
- To provide the engineering students with necessary background for understanding various nanomaterials characterization techniques
- To develop an understanding of the basis of the choice of material for device applications
- To give an insight into complete systems where nanotechnology can be used to improve our everyday life

Teaching-Learning Process

These are sample Strategies, which teacher can use to accelerate the attainment of the various course outcomes and make Teaching –Learning more effective

- 1. Chalk and Talk
- 2. Powerpoint presentation
- 3. Video Lecturing
- 4. E-sources
- 5. Self learning

Module-1 (07 hours of pedagogy)

Introduction to Nanomaterials

Nanotechnology, Frontier of future-an overview, Length Scales, Variation of physical properties from bulk to thin films to nanomaterials, Confinement of electron in 0D, 1D, 2D and 3D systems, Surface to Volume Ratio, Synthesis of Nanomaterials: Bottom-Up approach: Chemical Routes for Synthesis of nanomaterials-Sol-gel, Precipitation, Solution Combustion synthesis, Hydrothermal, SILAR, Chemical Bath Deposition. Top-Down approach- Ball milling technique, Sputtering, Laser Ablation

Module-2 (07 hours of pedagogy)

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Characterization of Nanomaterials

Basic principles and instrumentations of Electron Microscopy –Transmission Electron Microscope, Scanning Electron Microscope, Scanning Probes- Scanning Tunneling microscope, Atomic Force Microscope –different imaging modes, comparison of SEM and TEM, AFM and STM, AFM and SEM.

Basic principles of working of X-ray diffraction, derivation of Debye-Scherrer equation, numericals on Debye Scherrer equation, Optical Spectroscopy- Instrumentation and application of IR, UV/VIS (Band gap measurement)

Module-3(07 hours of pedagogy)

Carbon Based Materials

Introduction, Synthesis, Properties (electrical, Electronic and Mechanical), and Applications of Graphene, SWCNT, MWCNT, Fullerenes and other Carbon Materials: Carbon nanocomposites, nanofibres, nanodiscs, nanodiamonds.

Module-4(07 hours of pedagogy)

Nanotechnology in Energy storage and conversion

Solar cells: First generation, Second generation and third generation solar cells: Construction and working of Dye sensitized and Quantum dot sensitized solar cells.

Batteries: Nanotechnology in Lithium ion battery- working, Requirements of anodic and cathodic materials, classification based on ion storage mechanisms, limitations of graphite anodes, Advances in Cathodic materials, Anodic materials, Separators

Fuel Cells:Introduction, construction, working of fuel cells and nanotechnology in hydrogen storage and proton exchange membranes

Self study for lifelong learning:

Super capacitors: Introduction, construction and working of supercapacitor

Module-5 (07 hours of pedagogy)

Applications of Nanotechnology

Nanotech Applications and Recent Breakthroughs: Introduction, Significant Impact of Nanotechnology and Nanomaterial, Medicine and Healthcare Applications, Biological and Biochemical Applications (Nano biotechnology), Electronic Applications (Nano electronics), Computing Applications (Nano computers), Chemical Applications (Nano chemistry), Optical Applications (Nano photonics), Agriculture and Food Applications, Recent Major Breakthroughs in Nanotechnology.

Self study for lifelong learning:

Nano coatings (Photocatalysts) and super hydrophobic coatings (Lotus effect)

Course outcome (Course Skill Set)

At the end of the course the student will be able to:

C01	Demonstrate the synthesis of nanoparticles by various techniques. [L2]
CO2	Explain working of basic instruments used in characterization of nanoparticles. [L2]
CO3	Discuss the application of nanotechnology to mechanical and civil domains [L2]
CO4	Classify the nanomaterials based on the dimensions. [L3]
C05	Assess the suitability of nanomaterials for various device applications. [L4]

Assessment Details (both CIE and SEE)

The weightage of Continuous Internal Evaluation (CIE) is 50% and for Semester End Exam (SEE) is 50%. The minimum passing mark for the CIE is 40% of the maximum marks (20 marks out of 50). The minimum passing mark for the SEE is 35% of the maximum marks (18 marks out of 50). A student shall be deemed to have satisfied the academic requirements and earned the credits allotted to each subject/ course if the student secures not less than 35% (18 Marks out of 50) in the semester-end examination(SEE), and a minimum of 40% (40 marks out of 100) in the sum total of the CIE (Continuous Internal Evaluation) and SEE (Semester End Examination) taken together.

Continuous Internal Evaluation(CIE):

Three Tests each of 20 Marks;

- 1st, 2nd, and 3rd tests shall be conducted after completion of the syllabus of 30-35%, 70-75%, and 90-100% of the course/s respectively.
- Assignments/Seminar/quiz/group discussion /field survey & report presentation/ course project/Skill development activities, suitably planned to attain the COs and POs for a total of 40 Marks.

If the nature of the courses requires assignments/Seminars/Quizzes/group discussion two evaluation components shall be conducted. If course project/field survey/skill development activities etc then the evaluation method shall be one.

Total CIE marks (out of 100 marks) shall be scaled down to 50 marks

Semester End Examination (SEE):

Theory SEE will be conducted by University as per the scheduled timetable, with common question papers for the subject (**duration 03 hours**)

- The question paper shall be set for 100 marks. The medium of the question paper shall be English). The duration of SEE is 03 hours.
- The question paper will have 10 questions. Two questions per module. Each question is set for 20 marks. The students have to answer 5 full questions, selecting one full question from each module. The student has to answer for 100 marks and marks scored out of 100 shall be proportionally reduced to 50 marks.
- There will be 2 questions from each module. Each of the two questions under a module (with a maximum of 3 sub-questions), **should have a mix of topics** under that module.

Suggested Learning Resources:

Books (Title of the Book/Name of the author/Name of the publisher/Edition and Year)

- 1. Nano Materials A.K. Bandyopadhyay/ New Age Publishers
- 2. Nanocrystals: Synthesis, Properties and Applications C.N.R. Rao, P. John Thomas and G. U. Kulkarni, Springer Series in Materials Science
- 3. Nano Essentials- T. Pradeep/TMH
- 4. Peter J. F. Harris, Carbon nanotube science: synthesis, properties, and applications. Cambridge University Press, 2011
- 5. M.A. Shah, K.A. Shah, "Nanotechnology: The Science of Small", Wiley India, ISBN 13: 9788126538683

Reference Books (Title of the Book/Name of the author/Name of the publisher/Edition and Year)

- 1. Introduction to Nanotechnology, C. P. Poole and F. J. Owens, Wiley, 2003
- 2. Understanding Nanotechnology, Scientific American 2002
- 3. Nanotechnology, M. Ratner and D. Ratner, Prentice Hall 2003
- 4. Nanotechnology, M. Wildon, K. Kannagara, G. Smith, M. Simmons and B. Raguse, CRC Press Boca Raton 2002

5. Recent reviews on Li-ion batteries, solar cells and fuel cells

Web links and Video Lectures (e-Resources):

- https://nptel.ac.in/courses/118104008
- <u>https://www.digimat.in/nptel/courses/video/118104008/L16.html</u>
- https://archive.nptel.ac.in/courses/113/106/113106099/
- <u>https://nptel.ac.in/courses/112107283</u>
- <u>https://onlinecourses.nptel.ac.in/noc22_me131/preview</u>

Practical Based learning (Any 5 experiments x 2 hours = 10 practical hours)

- Preparation of silver nanoparticles and characterization of particle size by optical spectroscopy
- Preparation of ZnO nanoparticles by combustion technique
- Preparation of Al₂O₃ nanoparticles by precipitation method
- Preparation of Silica nanoparticles by sol-gel method
- Preparation of metal oxide nanoparticles by hydrothermal method
- Determination of thermal conductivity of nanofluids using a thermal analyser
- Preparation of thin films by SILAR method
- Determination of Band gap of given material using Tauc plot

COs and POs Mapping (Individual teacher has to fill up)

	POs										
1	2	3	4	5	6	7	8	9	10	11	12
3	3	2						2	1		
3	3	2									
3	3										
3	3							2	1		2
3	3							2	1		2
-	3 3 3 3 3	1 2 3 3 3 3 3 3 3 3 3 3 3 3 3 3	1 2 3 3 3 2 3 3 2 3 3 2 3 3 2 3 3 2 3 3 2 3 3 2	1 2 3 3 3 2 3 3 2 3 3 2 3 3 2 3 3 2 3 3 2	2 2 3 1 3 3 2 3 3 2 3 3 2 3 3 2 3 3 2	1 2 3 1 5 5 3 3 2	1 2 3 1 0 0 3 3 2	1 2 3 2 5 5 5 6 3 3 2 3 3 2 3 3 3 3 3 3	1 2 3 3 2 3 3 2 3 3 2 1 1 2 3 3 2 1 1 2 3 3 2 1 1 2 3 3 1 1 2 3 3 1 1 2 3 3 1 1 2 3 3 1 1 2	1 2 3 3 2 1 3 3 2 1 3 3 2 2 1 2 1 3 3 2 2 1 2 1 3 3 2 2 1 2 1 3 3 2 2 1 2 1 3 3 2 2 1 2 1 3 3 2 2 1 2 1	3 3 2 3 3 2 1 3 3 2 1 1 1 3 3 2 1 1 3 3 1 1 1 3 3 1 1 1

16-2-2023

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